

Supporting Information

On the Structure and Relative Stability of Uranyl(VI) Sulfate Complexes in Solution.

Valérie Vallet¹ and Ingmar Grenthe²

¹Laboratoire PhLAM (CNRS UMR 8523), CERLA (CNRS FR 2416), Université de Sciences et Technologies de Lille1, F-59655 Villeneuve d'Ascq Cedex, France

²Department of Chemistry, Inorganic Chemistry, Royal Institute of Technology (KTH), Teknikringen 36, S-10044 Stockholm, Sweden

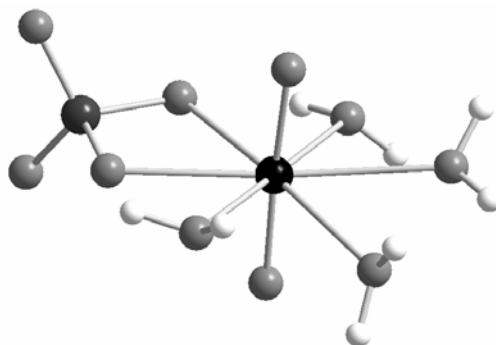


Figure S1. Structure of the six coordinated isomer $[\text{UO}_2(\text{SO}_{4\text{-chel}})(\text{OH}_2)_4]$. The average distance $\text{U} - \text{O}_{\text{eq}}$ is 2.51 Å. There are only very weak hydrogen bond interactions within the first coordination sphere with $\text{H} - \text{O}$ distances larger than 2.1 Å.

¹ valerie.vallet@univ-lille1.fr

² ingmarg@kth.se

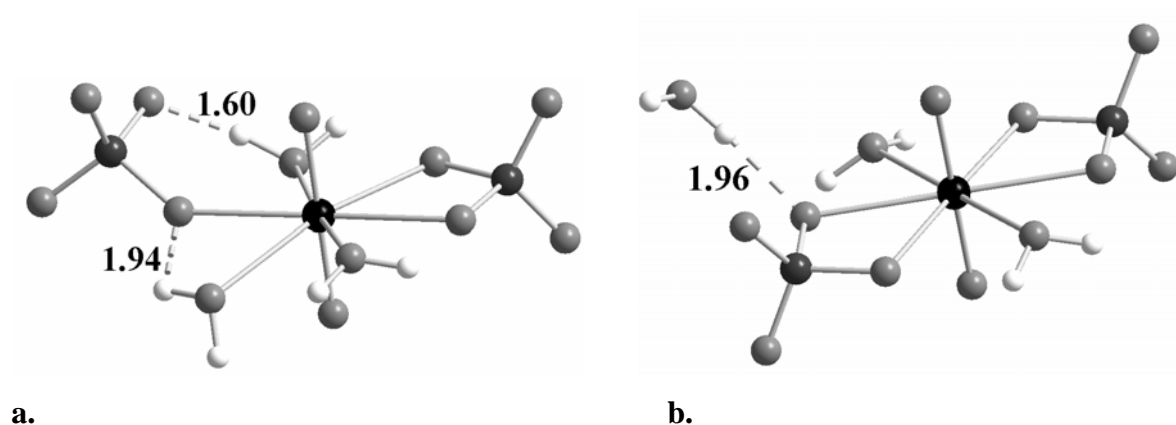


Figure S2. Structure of the six coordinated bonding isomer of $\text{UO}_2(\text{SO}_4)_2^{2-}$: a. $[\text{UO}_2(\text{SO}_4\text{-chel})(\text{SO}_4\text{-mono})(\text{OH}_2)_3]^{2-}$ and b. $[\text{UO}_2(\text{SO}_4\text{-chel})_2(\text{H}_2\text{O})_2]^{2-}, (\text{OH}_2)$. The dashed lines denote hydrogen bonds. In a. there is only one reasonably strong hydrogen bond at 1.60 Å; the hydrogen bond interactions of the other two water ligands are very weak. In b. the second sphere water is very weakly bonded to one sulfate group. There are no hydrogen-bond interactions involving the coordinated water.

Table S1. Total electronic energies in au for the various complexes computed at the DFT-B3PW91 and MP2 levels in the CPCM solvent model.

Complex	E(B3PW91)	G(B3PW91)	E(MP2)	G(MP2)
UO ₂ (SO ₄ -mono)(OH ₂) ₄	-1632.62016456	-1632.552129	-1629.3577438738	-1629.289708
[UO ₂ (SO ₄ -mono)(OH ₂) ₄],Na	-1794.82900804	-1794.829008	-1791.1874599669	-1791.123513
[UO ₂ (SO ₄ -chel)(OH ₂) ₃],(H ₂ O)	-1632.62483164	-1632.557076	-1629.3639421147	-1629.296187
[UO ₂ (SO ₄ -chel)(OH ₂) ₃],(H ₂ O),Na	-1794.82900804	-	-1791.1913741436	-
[UO ₂ (SO ₄ -mono) ₂ (OH ₂) ₃] ²⁻	-2255.53025205	-2255.473006	-2251.3634496638	-2251.306204
[UO ₂ (SO ₄ -chel)(SO ₄ -mono)(OH ₂) ₂] ²⁻ , (H ₂ O)	-2255.52910522	-	-2251.3631341250	-
[UO ₂ (SO ₄ -chel)(SO ₄ -mono)(OH ₂) ₃] ²⁻	-2255.52093766	-2255.461753	-2251.3592256576	-2251.300041
[UO ₂ (SO ₄ -chel) ₂ (OH ₂) ₂] ²⁻ , (H ₂ O) ₂	-2255.52426795	-2255.475267	-2251.3601842663	-2251.311183
[UO ₂ (SO ₄ -chel) ₂ (H ₂ O) ₂] ²⁻ , (OH ₂)	-2255.52911480	-2255.471836	-2251.3702361557	-2251.312957

Table S2. Cartesian coordinates in Å of the various isomers of $[\text{UO}_2(\text{H}_2\text{O})_5]^{2+}, (\text{H}_2\text{O})$, $\text{UO}_2(\text{SO}_4)$, and $\text{UO}_2(\text{SO}_4)_2^{2-}$ complexes. Geometries were optimized with DFT-B3PW91 in the CPCM solvent model.

Complex	Figure	Coordinates
$[\text{UO}_2(\text{OH}_2)_5]^{2+}, (\text{H}_2\text{O})$	-	U -0.224313 0.022279 0.095159
		O -0.508930 -0.044188 -1.626245
		O 1.791862 1.386900 -0.155043
		O -0.879253 -2.307357 0.200912
		O -0.042168 0.086177 1.830502
		O 1.752680 -1.415259 -0.051269
		O -2.615897 0.090107 0.408957
		O -0.794390 2.378962 0.002215
		H -1.393355 -2.771969 -0.486926
		H -0.939482 -2.840595 1.016720
		H 1.743944 -2.278346 -0.501510
		H 2.685593 -1.097982 -0.008505
		H -3.293145 0.035617 -0.291821
		H -3.082683 0.146232 1.264218
		H 1.811160 2.191705 -0.702690
		H 2.713861 1.041554 -0.082407
		H -1.366302 2.777540 -0.681423
		H -0.777104 2.995943 0.758789
		O 4.043229 -0.035257 0.080922
		H 4.683129 -0.069142 -0.653497
H 4.571185 -0.006650 0.899652		
$\text{UO}_2(\text{SO}_4\text{-mono})(\text{OH}_2)_4$	Fig. 1a	U -0.614795 -0.071315 -0.117260
		O -0.516294 0.017170 1.636413
		O -0.835550 -0.176963 -1.860933
		O -0.519688 2.419061 -0.267682
		O -2.903836 0.812099 -0.001099
		O -2.200276 -2.004901 0.131325
		O 0.555968 -2.200887 -0.046763
		O 1.537511 0.435197 -0.313979
		S 2.959346 0.096948 0.315456
		O 3.967510 0.392471 -0.734301
		O 2.886972 -1.388905 0.627215

		O	3.117718	0.910983	1.546997
		H	0.330708	2.839732	-0.487034
		H	-1.216816	2.970768	-0.664053
		H	-3.367587	1.109615	0.803602
		H	-3.504066	0.967136	-0.754169
		H	-3.155077	-1.901347	0.288243
		H	-1.923186	-2.795369	0.627890
		H	1.518323	-2.012917	0.217822
		H	0.576194	-2.773395	-0.833452
[UO ₂ (SO ₄ -mono)(OH ₂) ₄],Na	-	O	-0.205041	0.008773	-1.605295
		U	-0.614944	-0.086998	0.099812
		O	-2.196447	-1.877227	-0.488917
		O	-0.476423	2.381800	0.345837
		O	1.485180	0.409345	0.743812
		S	2.998545	0.081550	0.488696
		O	3.497578	0.950919	-0.621264
		O	-1.157312	-0.233236	1.765150
		O	0.582933	-2.208070	0.201666
		O	-2.753437	1.021299	-0.410723
		O	3.040699	-1.384879	0.138035
		O	3.744005	0.392839	1.747734
		H	1.574026	-2.045073	0.186221
		H	0.396227	-2.839803	0.917879
		H	-3.192847	0.992266	-1.280328
		H	-3.445265	1.204283	0.251278
		H	-1.014046	3.011611	-0.164252
		H	0.340561	2.838949	0.609449
		H	-2.795872	-2.316940	0.141443
		H	-2.035194	-2.507082	-1.214809
		Na	5.158316	2.113734	0.758463
[UO ₂ (SO ₄ -chel)(OH ₂) ₃],(H ₂ O)	Fig 1b	U	0.541201	0.091317	-0.038117
		O	0.530672	-0.337283	-1.744650
		O	-1.384467	1.443566	-0.242006
		O	-3.269880	0.745323	1.243235
		O	0.705659	0.477282	1.670809
		O	0.913559	-2.241724	0.489187

		O	2.968155	0.053731	-0.152325
		O	1.207785	2.424657	-0.613072
		O	-1.633964	-0.860782	0.245773
		O	-3.361480	0.244149	-1.191201
		O	-1.198760	-3.582396	-0.173405
		S	-2.522634	0.407473	0.014109
		H	0.183066	-2.875366	0.206538
		H	1.202099	-2.507277	1.379882
		H	3.505843	-0.221585	-0.917988
		H	3.583059	0.257741	0.576629
		H	0.520898	3.109651	-0.700937
		H	1.895638	2.635317	-1.269615
		H	-1.760999	-2.787870	-0.182529
		H	-1.288335	-3.990727	-1.049866
[UO ₂ (SO ₄ -chel)(OH ₂) ₃],(H ₂ O),Na	-	U	0.225281	0.152580	-0.011316
		O	0.174333	-0.450695	-1.663826
		O	-1.482125	1.778116	-0.423026
		O	-3.397903	1.566264	1.156588
		O	0.429154	0.713055	1.644394
		O	0.278882	-2.101212	0.767040
		O	2.621991	-0.309694	-0.065921
		O	1.282307	2.243002	-0.805611
		O	-2.066033	-0.382102	0.359460
		O	-3.650069	0.787318	-1.186238
		O	0.051120	-4.286531	-0.610150
		S	-2.741342	0.959116	-0.034551
		H	0.187945	-2.940635	0.220381
		H	0.278150	-2.348739	1.707578
		H	3.115320	-0.733245	-0.791281
		H	3.265272	-0.080216	0.628721
		H	0.964400	3.111546	-0.499680
		H	1.558876	2.360922	-1.732418
		H	-0.725176	-4.355551	-1.190680
		H	0.812714	-4.528736	-1.163103
		Na	-5.368287	2.372707	1.925872
[UO ₂ (SO ₄ -chel)(OH ₂) ₄]	Fig. S1	U	0.485642	0.129034	-0.050409

		H	1.505511	-2.115950	-0.556545
		H	-2.031680	2.697607	-0.386858
		H	-2.617857	1.546069	0.542545
		H	0.681870	3.513938	0.244339
		H	1.967298	2.678476	0.543980
[UO ₂ (SO ₄ -chel)(SO ₄ -mono)(OH ₂) ₂] ²⁻ ,(H ₂ O)	Fig. 2b	U	0.039985	-0.378812	-0.098205
		O	-0.148675	-0.532458	1.653189
		O	0.222336	-0.291631	-1.856758
		O	-2.206160	-0.252470	-0.364231
		S	-3.529833	0.441650	0.128143
		O	-3.160074	1.905712	0.309189
		O	-4.543736	0.267608	-0.950188
		O	-3.939518	-0.181541	1.418250
		O	-0.579231	2.001442	0.060257
		H	-0.171861	2.537802	0.761022
		H	-1.577574	2.083634	0.166092
		O	-0.394007	-2.784710	-0.318918
		H	-0.853019	-3.342495	0.336195
		H	-0.347628	-3.294408	-1.148910
		O	2.042370	0.968172	0.196826
		S	3.126389	-0.154472	0.237743
		O	3.802450	-0.172506	1.558324
		O	2.221383	-1.401463	0.063411
		O	4.060934	-0.017371	-0.905854
		H	2.503041	2.760212	-0.156040
O	2.710391	3.699647	-0.305345		
H	2.515862	3.843643	-1.242840		
[UO ₂ (SO ₄ -chel)(SO ₄ -mono)(OH ₂) ₃] ²⁻	Fig. S2a	U	0.110179	-0.262988	-0.080712
		O	0.354733	-0.237565	-1.830474
		O	-0.129219	-0.293526	1.666318
		O	-2.261148	-0.267958	-0.284553
		S	-3.552530	0.283117	0.413138
		O	-3.277677	1.752018	0.696922
		O	-4.674675	0.126443	-0.561246
		O	-3.782562	-0.476275	1.677380
		O	2.558249	-0.455714	0.202560

		S	2.972528	1.028778	0.324521
		O	1.581119	1.691951	0.194441
		O	3.569298	1.303561	1.658623
		O	3.858396	1.429660	-0.800400
		O	-1.160599	-2.513470	-0.729515
		H	-0.922003	-2.915512	-1.584158
		H	-2.036723	-2.109558	-0.852372
		O	1.225121	-2.554407	0.288549
		H	0.898181	-3.429464	0.026225
		H	2.193959	-2.562848	0.236404
		O	-0.881863	2.010241	-0.246257
		H	-0.343526	2.709955	0.155383
		H	-1.807268	2.052511	0.135792
$[\text{UO}_2(\text{SO}_4\text{-chel})_2(\text{H}_2\text{O})]^{2-}, (\text{OH}_2)_2$	Fig. 2c	U	-0.003093	-0.001188	0.276154
		O	-1.731691	0.354795	0.381089
		O	-0.339800	-1.532170	-1.547164
		S	-0.620213	-2.849920	-0.752809
		O	-2.000228	-3.323215	-1.014512
		O	0.067077	-0.001306	2.711813
		O	1.728652	-0.357507	0.289355
		O	0.502515	2.312335	0.676994
		S	0.563516	2.848498	-0.778983
		O	-0.493234	3.860955	-1.011260
		O	-0.486051	-2.315655	0.698984
		O	0.239756	1.532410	-1.559863
		O	1.929375	3.318982	-1.110474
		O	0.425603	-3.859518	-1.040788
		H	-0.682666	0.168988	3.312991
		H	0.858433	-0.138971	3.266144
		H	0.718541	-1.693629	-3.133834
		O	1.266291	-1.818497	-3.927727
		H	1.834672	-2.570871	-3.708148
		H	-0.910031	1.702751	-3.082493
		O	-1.498186	1.832926	-3.845958
		H	-2.086454	2.551760	-3.573020
$[\text{UO}_2(\text{SO}_4\text{-chel})_2(\text{H}_2\text{O})]^{2-}, (\text{OH}_2)$	Fig.	U	-0.071119	0.054091	0.136465

S2b	O	1.654400	0.276557	-0.180555
	O	-1.797784	-0.167634	0.449941
	O	-0.046862	-1.938912	-1.296029
	S	0.217492	-3.056327	-0.257531
	O	0.315137	-2.201854	1.028958
	O	1.507527	-3.741517	-0.535501
	O	-0.934126	-3.994012	-0.185568
	O	-0.448307	2.342992	-0.744661
	S	-0.425227	3.168620	0.572230
	O	-0.168410	2.032537	1.587915
	O	0.701402	4.137025	0.560924
	O	-1.744419	3.808576	0.813180
	O	0.545762	-0.197407	2.583805
	H	0.227724	0.485001	3.198251
	H	0.390853	-1.064872	2.993965
	O	-0.653268	0.315968	-2.318735
	H	-0.992538	-0.491802	-2.741098
	H	-1.258722	1.041125	-2.549316
	H	0.541756	3.141032	-2.176228
	O	1.001349	3.599478	-2.901426
	H	1.373931	4.389345	-2.484028
