

-Supplementary data-

Étude structurale des analogues du bleu de Prusse par spectroscopie infra-rouge

Application of the infra-red spectroscopy to the structural study of Prussian blue analogues

Julien Lejeune^{a,*}, Jean-Blaise Brubach^b, Pascale Roy^{a,b}, Anne Bleuzen^{a,*}

^a *Institut de Chimie Moléculaire et des Matériaux d'Orsay, UMR CNRS 8182, Université Paris-Sud, 15
rue Georges Clémenceau, F-91405 Orsay Cedex, France*

^b *Synchrotron SOLEIL, l'Orme des merisiers Saint-Aubin, BP 48, F-91192 Gif-sur-Yvette Cedex, France*

-Supplementary data-

A1. Evolution of the different contributions to the $\nu\{\text{O-H}\}$ bands as a function of the chemical composition of A_xCoFe PBAs

The IR spectrum of PBAs exhibits very sharp bands in the $3550\text{-}3700\text{ cm}^{-1}$ spectral range, associated with water molecules that are not involved in a hydrogen-bonded network. The number of these sharp bands may vary depending on the nature of the PBA (Fig. A1).

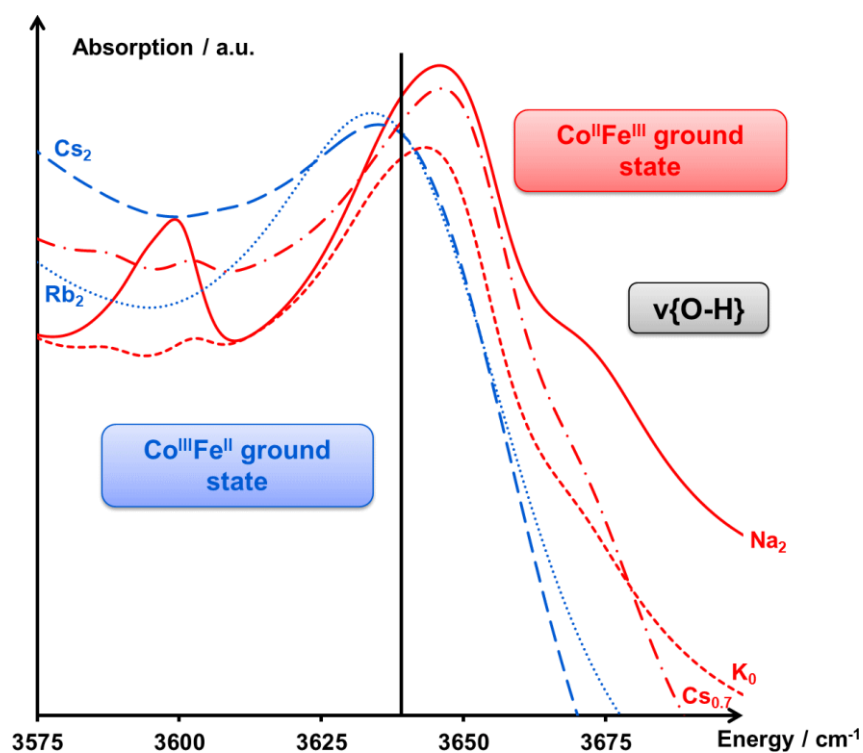


Fig. A1. IR spectrum of several A_xCoFe PBAs in the spectral range associated with the $\nu\{\text{O-H}\}$ vibration band of water molecules bound to the cobalt cations at the $\text{Fe}(\text{CN})_6$ vacancies, at $T = 300\text{ K}$. Spectra associated with A_xCoFe PBAs exhibiting a $\text{Co}^{\text{II}}\text{Fe}^{\text{III}}$ (resp. $\text{Co}^{\text{III}}\text{Fe}^{\text{II}}$) ground state at $T = 300\text{ K}$ are plotted in red (resp. blue). Legend: K_0CoFe (---), $\text{Cs}_{0.7}\text{CoFe}$ (-·-·-), Na_2CoFe (—), Rb_2CoFe (···), Cs_2CoFe (— — —).

The position in term of energy of the main band, associated with the $\nu\{\text{O-H}\}$ vibration of water molecules bound to the cobalt cations at the $\text{Fe}(\text{CN})_6$ vacancies, mostly depends on the electronic state of the system (and, therefore, of the oxidation state of the cobalt cations): in the

-Supplementary data-

Co^{II}Fe^{III} electronic state, the $\nu\{\text{O-H}\}$ vibration band associated with water molecules bound to the Co^{II} cations is located above 3638 cm^{-1} ; in the Co^{III}Fe^{II} electronic state, the $\nu\{\text{O-H}\}$ vibration band associated with water molecules bound to the Co^{III} cations is located below 3638 cm^{-1} . The nature and the amount of alkali cations inserted in the structure may modulate the position in terms of energy of this band.