

Electrochemical properties of the $[\text{SiW}_{10}\text{O}_{36}(\text{M}_2\text{O}_2\text{E}_2)]^{6-}$ Polyoxometalates series (M = Mo(V) or W(V); E = S or O) in aqueous medium: application to the electro-analysis of iodates.

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Supporting Information

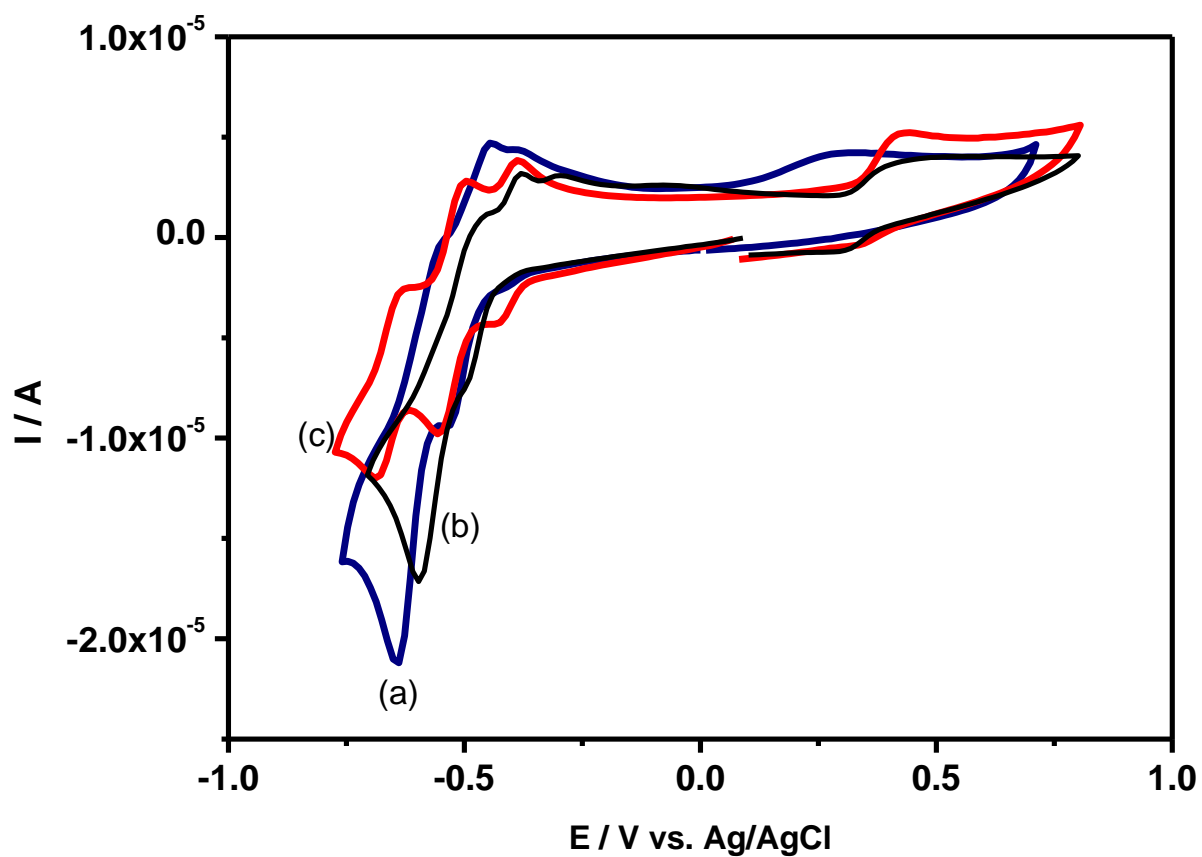


Figure S1: Cyclic voltammograms recorded at a glassy carbon electrode ($v = 0.1 \text{ V s}^{-1}$) in $0.1 \text{ M H}_2\text{SO}_4/\text{Li}_2\text{SO}_4$ at $\text{pH} = 1.5$ of solutions containing $0.2 \text{ mM}\{\text{SiW}_{10}\text{-Mo}_2\text{O}_4\}$ (a), $\{\text{SiW}_{10}\text{-Mo}_2\text{O}_2\text{S}_2\}$ (b) and $\{\text{SiW}_{10}\text{-W}_2\text{O}_2\text{S}_2\}$ (c),

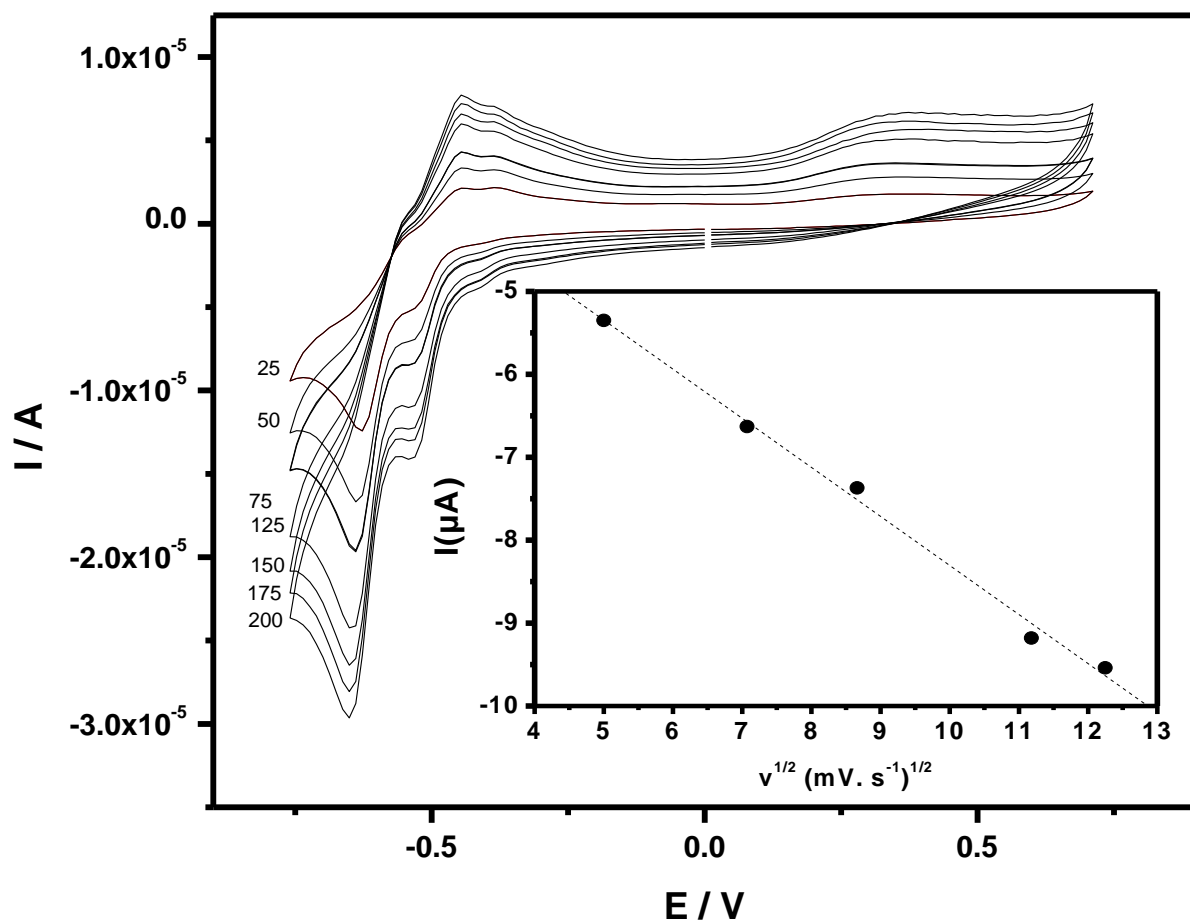


Figure S2. Cyclic voltammetry of 0.2 mM $\{\text{SiW}_{10}\text{-Mo}_2\text{O}_2\text{S}_2\}$ recorded at a glassy carbon electrode in 0.1 M $\text{H}_2\text{SO}_4/\text{Li}_2\text{SO}_4$ (pH = 1.5) at various scan rate ($\nu = 25 - 200 \text{ mVs}^{-1}$). Inset: variation of I_{pc} of the first reduction versus $\nu^{1/2}$ indicating a diffusion controlled process.

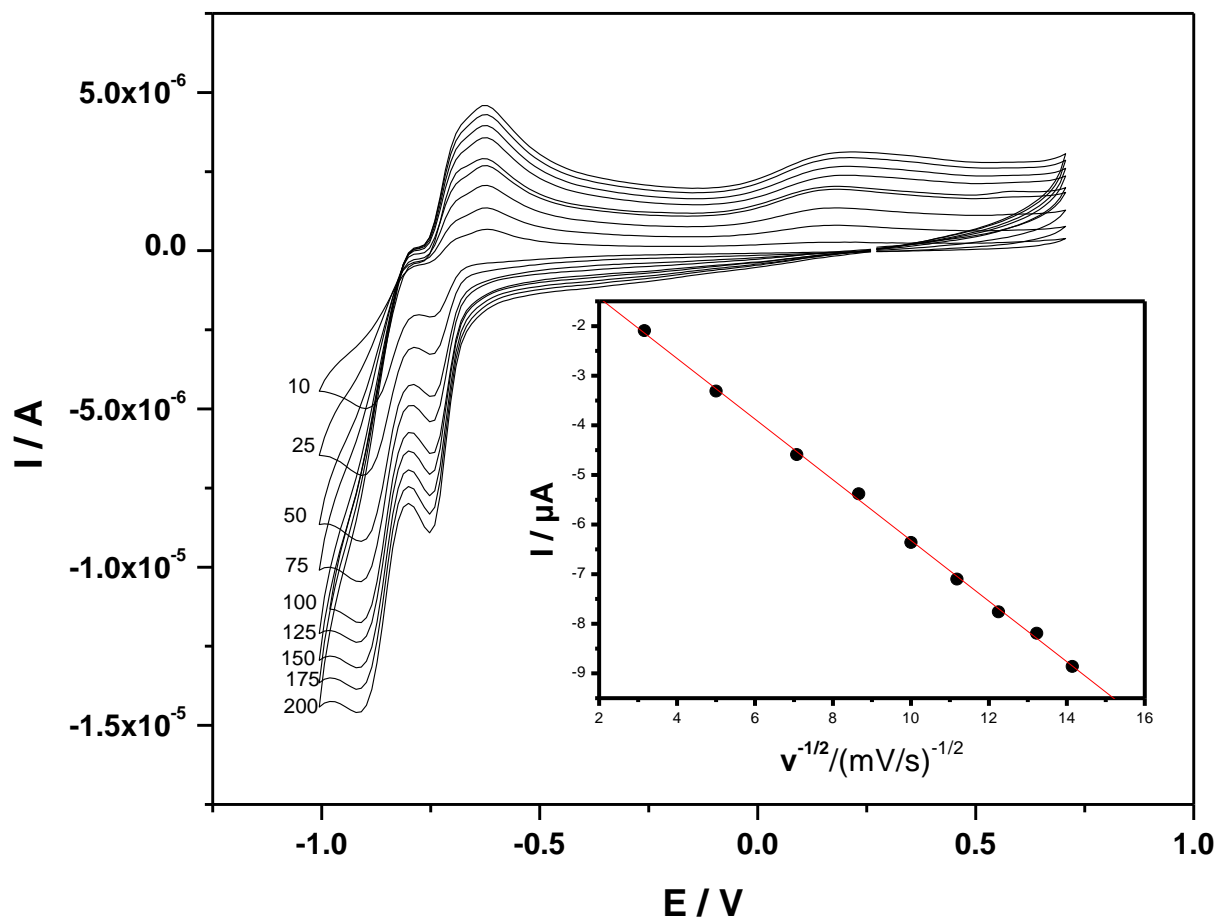


Figure S3: Cyclic voltammetry of 0.2 Mm {SiW₁₀-Mo₂O₄} recorded at a glassy carbon electrode in 0.1 M H₂SO₄/Li₂SO₄ (pH = 2) at various scan rate ($v = 0.1 - 2 \text{ Vs}^{-1}$). Inset: I_{pc} of the first reduction versus $v^{1/2}$.

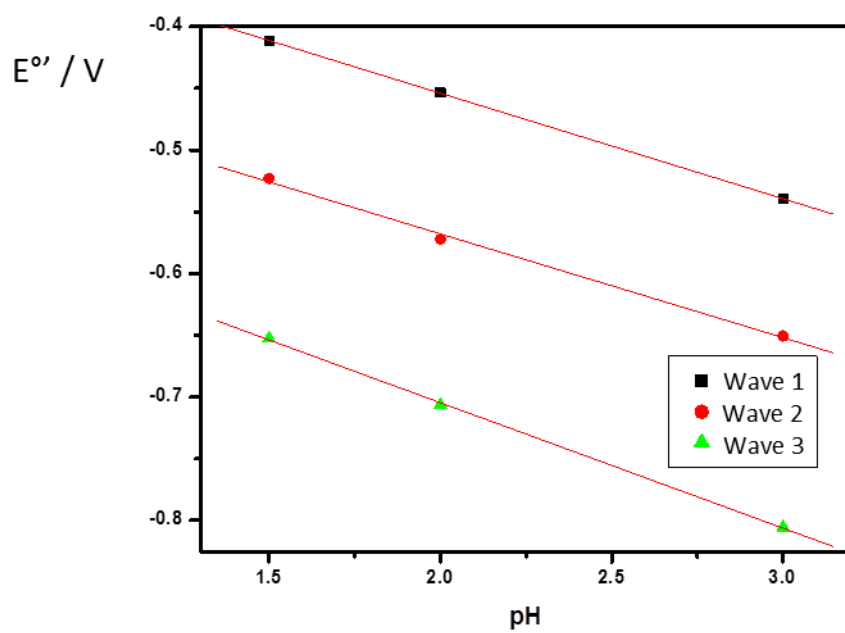


Figure S4.Dependence of $E^{\circ'}$ versus the buffer pH for $\{\text{SiW}_{10}\text{-W}_2\text{O}_2\text{S}_2\}$.