



Account/Revue

# Bacteria love our sugars: Interaction between soluble lectins and human fucosylated glycans, structures, thermodynamics and design of competing glycoconjugates

Aymeric Audfray, Annabelle Varrot, Anne Imberty\*

CERMAV-CNRS (affiliated to Université Joseph-Fourier and ICMG), 601, rue de la Chimie, BP 53, 38041 Grenoble cedex 9, France

## ARTICLE INFO

## Article history:

Received 6 September 2012

Accepted after revision 29 November 2012

Available online 12 January 2013

## Keywords:

Lectins

Glycoconjugates

Crystal structure

Glychochemistry

Fucose

## ABSTRACT

The specific recognition of host epithelial surfaces by pathogen receptors is often the first step of infection. Bacterial carbohydrate-binding proteins, called adhesins or lectins, specifically recognize glycoconjugates on host tissues. Opportunistic pathogens responsible for lung infections, such as some *Pseudomonas* and *Burkholderia* species, produce soluble lectins that bind to human fucosylated oligosaccharides. Among these targets, the observed variations in ABO and Lewis histo-blood group phenotypes are proposed to be a result of co-evolution between pathogens and mammals. The analysis of the available crystal structures of bacterial lectins helps deciphering the structure/function relationship involving this important class of protein. The high affinity observed between bacterial lectins and fucosylated oligosaccharides can be reached by different strategies in term of structures and thermodynamics. Structure-based design of glycomimetics and glycodendrimers resulted in very high affinity ligands that open the route towards novel anti-infectious strategies.

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## 1. Introduction

Lectins are proteins of non-immune origin that bind in specific and reversible ways to carbohydrates without modifying them [1,2] and often display the property of agglutinating cells. Lectins are widely distributed in nature and they have been identified in all part of living reign, from microorganisms to fungi, plants and animals. Due to their interactions with glycoproteins, glycolipids and oligosaccharides, lectins are involved in cell-cell communication and they play the crucial role of deciphering the glycode [3]. Their biological functions are very diverse. Lectins are important players in the innate immunity system of many invertebrates since they are able to differentiate between self and non-self carbohydrates [4]. Lectins are particularly abundant in plants, where they are proposed to be involved

in protection against pathogens or feeders [5]. The biological roles of animal lectins have been the most studied; they include among others, glycoprotein trafficking and clearance, development, immune defence, fertility and others [6].

In pathogenic microorganisms, lectins are often involved in host recognition and tissue adhesion [7]. Indeed, all known organisms are covered with arrays of free or covalently attached glycans that interact with receptors, cells and microorganisms. These glycans present a wide range of structural diversity, correlated with variability of possible configurations of monosaccharides and variability of linkages between monosaccharide units, including branching points [8]. The infection strategies of pathogens involve interactions characterized by their high specificity to host glycans and most of the time by multivalency. Interfering with synthetic compounds that mimic the natural oligosaccharides and effectively compete with the attachment site is an attractive anti-infectious strategy [9] that is now developed in many laboratories of glycochemistry.

\* Corresponding author.

E-mail address: [imberty@cermav.cnrs.fr](mailto:imberty@cermav.cnrs.fr) (A. Imberty).

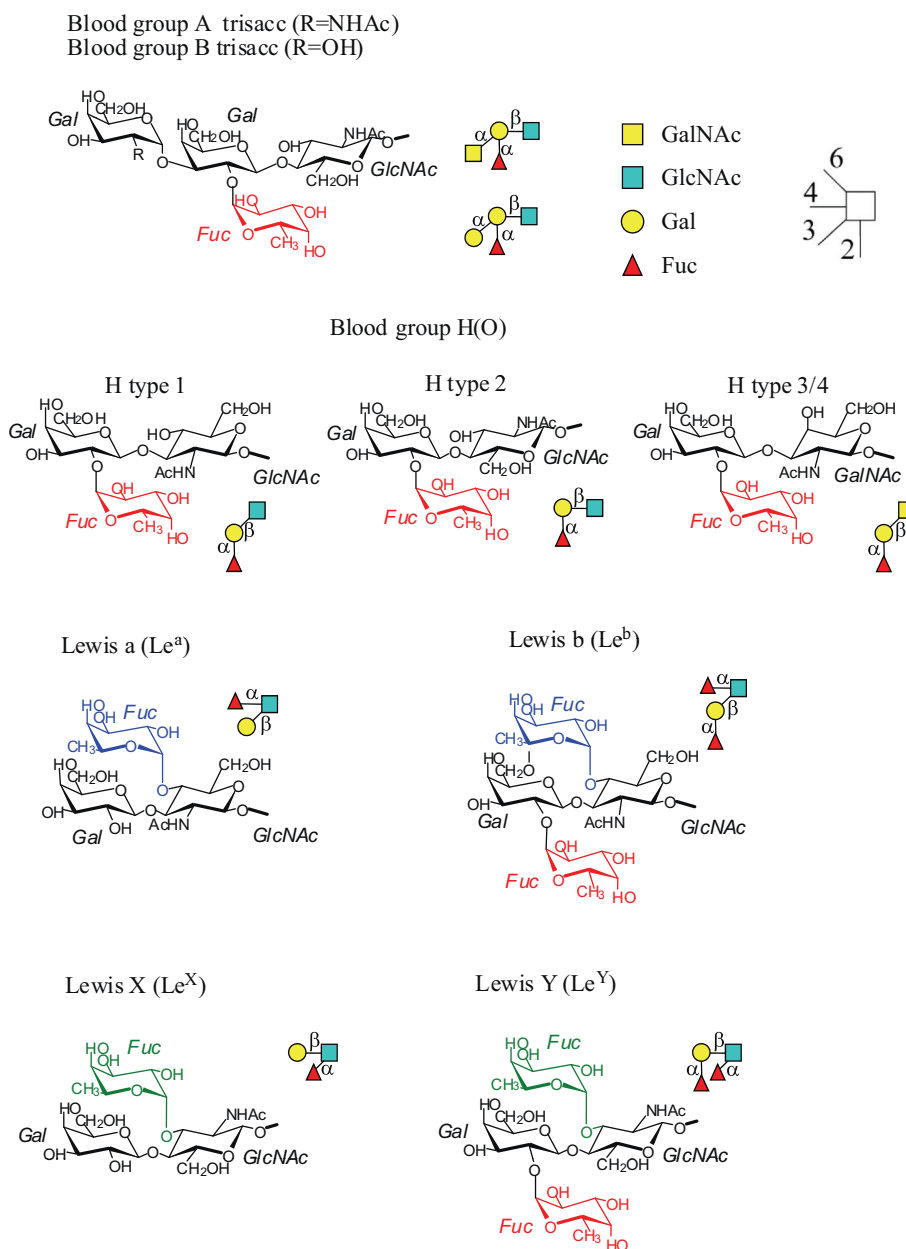


Fig. 1. Fucosylated histo-blood group oligosaccharides of interest with fucose residues colored in red when  $\alpha$ 1-2 linked to galactose, and in green and blue when  $\alpha$ 1-3 and  $\alpha$ 1-4 linked to GlcNAc, respectively. The schematic international representation is also given.

The present review focuses on the interactions occurring between opportunistic lung pathogens and their host as mediated by soluble lectin and glycoconjugates. The involvement of fucosylated histo-blood group oligosaccharides is discussed as well as the structural basis for the interaction. The knowledge accumulated in the recent years is now utilized for structure-based design of high affinity fucosylated compounds that can compete with the binding of receptors and therefore act as anti-adhesive compounds that can play a role in anti-infectious strategy.

## 2. Pathogen receptors and human glycans – an evolutionary point of view

Pathogenic organisms such as viruses, bacteria, fungi or parasites express different glycan binding proteins such as toxins, soluble lectins, and lectin domains attached for example on the tip of bacterial pili and flagella or to the surface of viral capsids and envelopes [7,10–12]. Evolution that resulted in this variety of receptors has been mirrored by selection of glycans on the host tissue. Although, in the absence of fossils, it is not

**Table 1**

Different families of bacterial soluble lectins from opportunistic pathogens and their specificity towards oligosaccharides.

Bacteria	Lectin	Higher affinity ligand	Oligomer (number of sites)	Reference
<i>Pseudomonas aeruginosa</i>	PA-IL (LecA)	$\alpha$ Gal4 Gal	Tetramer	[58]
<i>Pseudomonas aeruginosa</i>	PA-III (LecB)	H-type 2/Lewis a	Tetramer	[26]
<i>Chromobacterium violaceum</i>	CV-III	H-type 2	Tetramer	[24]
<i>Burkholderia cenocepacia</i>	BC2L-A	Oligomannose	Dimer	[59]
<i>Burkholderia cenocepacia</i>	BC2L-C	H-type 1/Lewis B	Trimer	[27]
	N-ter dom			
<i>Burkholderia ambifaria</i>	BamBL	H-type 2/Lewis Y	Trimer (6 sites)	[25]

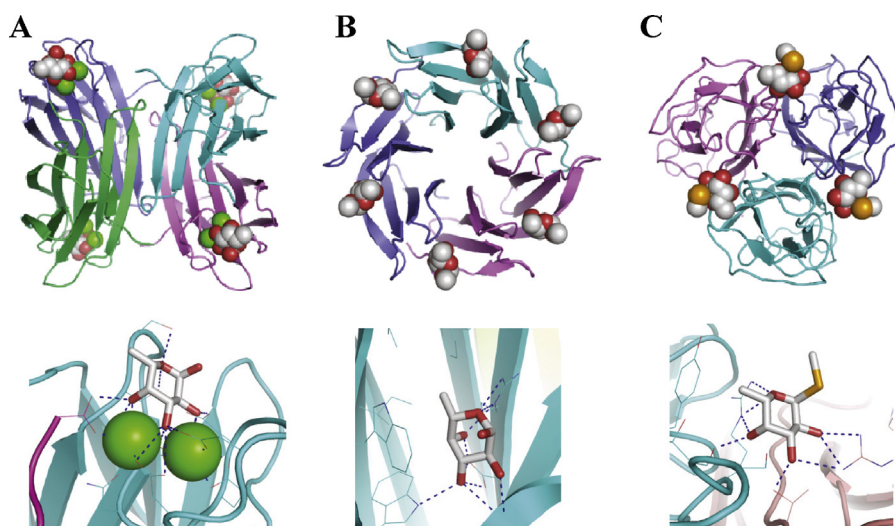
possible to obtain information from extinct pathogens, it is now accepted that a significant portion of the overall diversity in vertebrate cell-surface glycan structure reflects pathogen-mediated selection processes [13]. The glycans that are present on epithelia as part of mucins, N-glycoproteins or glycolipids can be the target for microbial adhesion, providing convenient points of invasion. However, they can also act as protection against pathogens, and mucins secretions have been shown to adsorb and trap intruding pathogens before they can invade the underlying tissues [14].

In this context, the individual variation of glycans present at the surface of human tissues reflects the pressure of past epidemic or endemic pathogens. Such variations are observed in blood group phenotypes since A, B and O blood groups and Lewis histo-blood groups (Fig. 1) are distinguished by oligosaccharides on the surface of cells. The A, B, and H(O) antigens are complex fucosylated oligosaccharides present on endothelial cells and erythrocytes of all individuals [15]. The ABH antigens are also expressed in saliva, tears, and mucus secretions in the digestive tract of individuals who display the secretor phenotype (80% of the Caucasian population). The biological role of the ABO blood group system remains mysterious but in the 1980s, anthropological surveys suggested that the geographic and racial distribution of

human blood groups reflects susceptibility of populations with specific blood types to the plague, cholera, smallpox, malaria and other infectious diseases [16,17]. The biochemical and structural evidences for the direct role of pathogen receptors interaction with ABH(O) oligosaccharides have been however obtained only very recently. Cholera toxin of the El Tor biotype of *Vibrio cholera* is captured by A and B blood group before entering epithelial cells, resulting in more severe infections for people with blood group O [18,19]. The ABO group also influences susceptibility to severe *Plasmodium falciparum* malaria due to an A/B oligosaccharides specific adhesin domains that promote rosetting, i.e. transmission of parasites from infected red cells to non-infected ones [20]. Susceptibility to viral gastroenteritis is also ABO phenotype-dependant, with variations of specificity between rotaviruses and the different strains of noroviruses [21].

### 3. Lectins from lung pathogens – convergence for binding to fucose

As described above, the histo-blood group correlation with susceptibility of infection has mainly been characterized for pathogens colonizing blood and gastro-intestinal tract. Lung pathogens are in contact with epithelia and mucins that are potential adhesion points. In the recent



**Fig. 2.** Crystal structure of three bacterial lectins complexed with fucose or fucoside with overview of the oligomeric state with ligand represented as spheres (top) and details of the binding site with ligand represented as stick (bottom) A. LecB/fucose (PDB code 1GZT) [23]. B. BamBL/fucose (PDB code 3ZWO) [25]. C. BC2L-C-nt/Me-Se-fucoside (PDB code 2WQ4).[27] Graphics have been prepared with PyMol program.

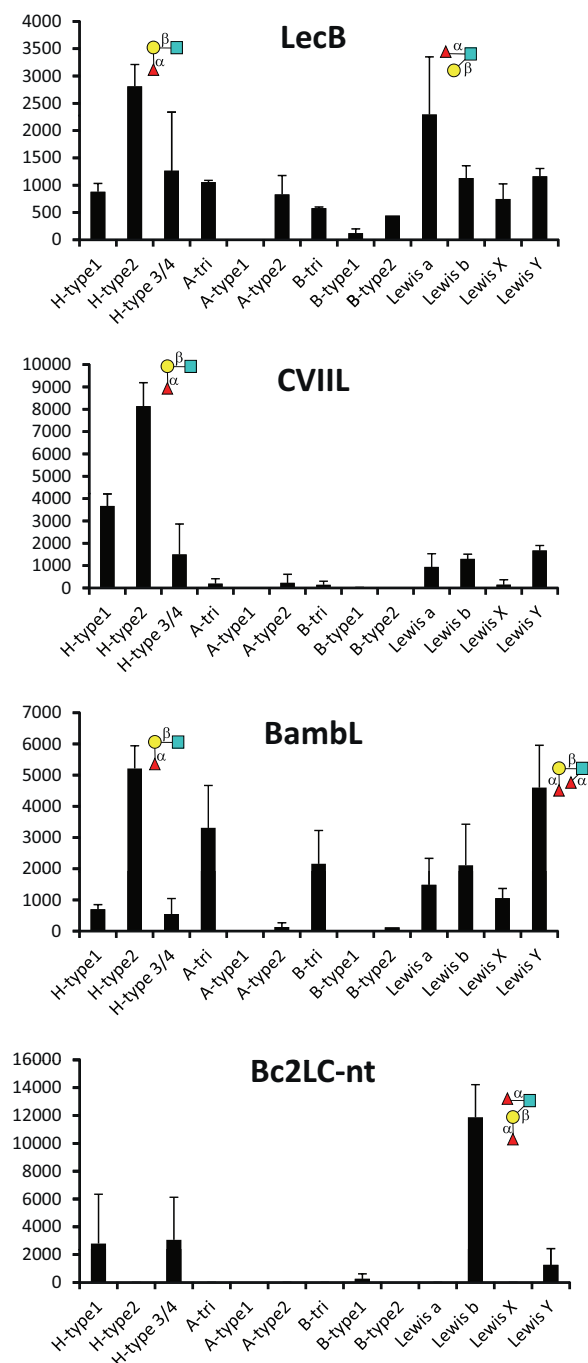


Fig. 3. Selected data for binding of Alexa fluor-labelled lectin to glycan array. Fluorescence intensities have been averaged over the different oligosaccharides that present the indicated epitopes in terminal position in a single copy. Complete glycan array data are available from <http://www.functionalglycomics.org>.

years, soluble lectins from opportunistic bacteria responsible for acute pneumonia in cystic fibrosis patients or in hospital environment have been described (Table 1).

Interestingly, most of these soluble lectins are fucose-specific and exhibit strong affinity towards human

fucosylated ABO or Lewis oligosaccharides. Three different families of soluble fucose-binding bacterial lectins have been identified in *Pseudomonas aeruginosa* and bacteria from the *Burkholderia cepacia* complex such as *Burkholderia cenocepacia* and *Burkholderia ambifaria* (Fig. 2). LecB from *P. aeruginosa* is a tetrameric protein that displays an unusually strong micromolar affinity to L-fucose in a tight binding site which requires two  $\text{Ca}^{2+}$  ions [22,23]. CV-III from the tropical pathogen *Chromobacterium violaceum* has strong sequence and structure similarities to LecB [24]. BamBL from *B. ambifaria* is a trimeric lectin arranged in a  $\beta$ -propeller fold with two similar binding sites per monomer, resulting in an hexameric arrangement of fucose-binding sites [25]. Previous specificity studies indicate that both lectins bind to a large variety of fucosylated oligosaccharides, with LecB having higher affinity to Lewis a epitope [26] and BamBL to H-type 2 epitopes [25]. A third fucose-binding bacterial lectin of interest is the N-terminal domain of Bc2L-C from *B. cenocepacia*. Bc2L-C is an hexameric lectin, each of the monomer containing two domains with different specificity and oligomerisation state [27,28]. The N-terminal domain assembles as a TNF $\alpha$ -like trimer with strong specificity for H-type1 and Lewis b oligosaccharides [27] while the C-terminal domain is a dimeric mannose-binding lectin [28].

Although all of the herein described bacterial lectins recognize fucose, their binding modes present strong differences. In LecB, as well as in the related CV-III, the specificity towards fucose is due to the presence of two calcium ions that bridges the protein acidic amino acids and the carbohydrate hydroxyl groups. Only monosaccharides presenting hydroxyl groups suite having axial-equatorial-equatorial arrangement, i.e. D-manno or L-galacto configuration, can fit in the coordination scheme. Fucose, that has the requested L-galacto configuration and a methyl group at C6, is the highest affinity ligand but other monosaccharides with this configuration such as L-galactose, D-mannose or D-arabinose can still bind to LecB and CV-III, albeit with lower affinity [24,29]. The fucose-binding sites of BamBL and Bc2L-C-nt are again different but they display one common feature. In both cases, an arginine residue is located directly under the fucose ring, establishing bifurcated hydrogen bonds with both the ring oxygen, and the axial O4 hydroxyl group, therefore imposing the specificity for L-galacto configuration.

#### 4. Fine specificities of fucose-binding lectins from lung pathogens

The fine specificity of lectins can be assessed with the use of “glycochips”, i.e. glycan array with more than 500 different human glycans that are available from the “Consortium for Functional Glycomics”. The four fucose-binding lectins of interest above have been produced recombinantly in *Escherichia coli*, then chemically labeled with Alexa Fluor and tested on these glycan arrays. The complete results are available from the Consortium web site (<http://www.functionalglycomics.org>). Fig. 3 summarizes their binding to ABO and Lewis epitopes. The structurally related lectins LecB and CV-III both bind to blood group O epitope H-type 2, although LecB also binds to

**Table 2**

Isothermal titration microcalorimetry data for the interaction between lectins of interest and the ligands fucose,  $\alpha$ -methyl-fucoside ( $\alpha$ MeFuc) and fucosylated oligosaccharides.

Lectin	Ligand	<i>n</i>	<i>K<sub>a</sub></i> ( $\mu$ M)	$-\Delta G$ (kJ/mol)	$-\Delta H$ (kJ/mol)	$T\Delta S$ (kJ/mol)	Reference
LecB	Fuc	0.96	2.9	31.5	31.2	0.3	[26]
	$\alpha$ MeFuc	1.05	0.38	36.6	35.5	1.1	[41]
	Lewis a	1.08	0.21	38.1	34.9	3.1	[26]
	Lewis x	0.95	3.4	31.1	22.3	8.8	[26]
CV-III	$\alpha$ MeFuc	0.95	1.68	32.9	21.8	11.1	[24]
BambL	$\alpha$ MeFuc	2.02	0.96	34.3	47.8	-13.5	[25]
	A-tri	1.96	0.46	36.0	53.1	-16.9	[25]
	H-type 2	1.96	7.5	29.3	44.4	-15.2	[25]
	Lewis Y	2.00	11.1	28.3	38.6	-10.3	[25]
BC2L-C-nt	$\alpha$ MeFuc	1 <sup>a</sup>	1236	14.7	20.9	-6.2	[27]
	H-type 1	0.93	77.2	23.5	23	0.5	[27]
	Lewis Y	0.98	53.9	24.4	34.9	-10.5	[27]
	Lewis b	1.11	213	21.0	19.8	1.15	[27]

<sup>a</sup> Value fixed during the fitting procedure.

Lewis a. BambL has a broader specificity and binds to blood group A, B, O and to Lewis oligosaccharides. On the opposite, BC2L-C-nt is highly specific for Lewis b tetrasaccharide.

### 5. Bacterial strategy for attaining high affinity – a thermodynamic point of view

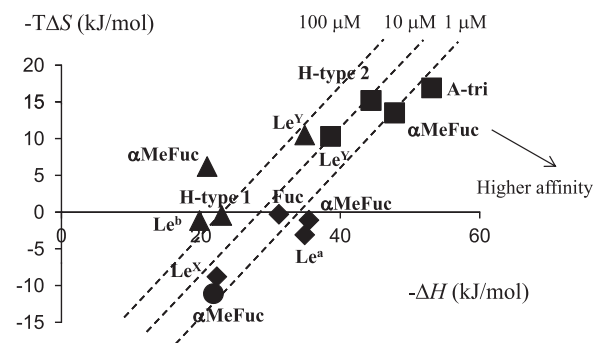
Titration calorimetry is a method of choice for quantification of protein/carbohydrate interactions. The large number of hydrogen bonds involved results in strongly exothermic interactions that are easily measured by isothermal titration calorimetry (ITC). Furthermore, the range of affinity, with dissociation constants generally between  $10^{-4}$  to  $10^{-7}$  M, is well suited for this method. Microcalorimetry experiments allow direct determination of affinity (association or dissociation constant) and stoichiometry of binding, and also of the contribution of enthalpy and entropy of binding.

Lectin-carbohydrate interactions are generally characterized by a favorable enthalpy term, due to the large number of hydrogen bonds, that is offset by an unfavorable entropy contribution that has been attributed either to solvent rearrangement or to loss of ligand conformational flexibility [30]. Our work on soluble bacterial lectins has

been accompanied by thermodynamic characterization that identified much higher affinity than that observed for plant or animal lectins (Table 2) with dissociation constants lower than micromolar for LecB and BambL for example. The detailed analysis of the binding energy data provides a way to evaluate the role played by the so-called enthalpy-entropy compensation that characterizes the interaction between protein and carbohydrate in solution. The analysis of the soluble bacterial lectins (Table 2 and Fig. 4) shows that only BambL follows these trends with a large favorable enthalpy of binding compensated by large unfavorable entropy of binding. LecB and BC2L-C-nt present slightly favorable entropy of binding. For LecB, this has been attributed to the unique involvement of two calcium ions in the binding site: In the native structure [31], three water molecules are tightly bound to the calcium, and binding of sugar releases them in the bulk of the solvent, resulting in a gain in entropy. This gain is even larger when binding to Lewis trisaccharides since these epitopes are branched oligosaccharides with very limited conformational variations in solution. CV-III, that also includes two calcium ions in the binding site, displays the stronger favorable entropy contribution [24]. Bacterial lectins have therefore been demonstrated to present higher affinity to glycan targets than classically observed, and to be able to use unusual entropy-driven strategies for reaching strong binding [32].

### 6. Structure-based design of carbohydrate-derived anti-infectious compounds

The LecB lectin from *P. aeruginosa* is an attractive target for anti-infectious developments since it is involved in biofilm formation [33], a phase of bacterial development that is specially dangerous in lung infection for cystic fibrosis patients. Recent data on related BC2L-A from *B. cenocepacia* points to possible role in formation of microcolonies by binding to heptose residues present on bacterial cell wall [34]. LecB has also been demonstrated to adhere to human airway cilia in respiratory cell culture and to lower the ciliary beat frequency, an effect that can be inhibited by fucose [35]. *P. aeruginosa* strain lacking LecB



**Fig. 4.** Enthalpy/entropy plot for the interaction of bacterial lectins with fucose or fucosylated oligosaccharides. Code for lectins is as follows: LecB: •; CV-III: ◼; BambL: ◼; BC2L-C-nt: ◼.

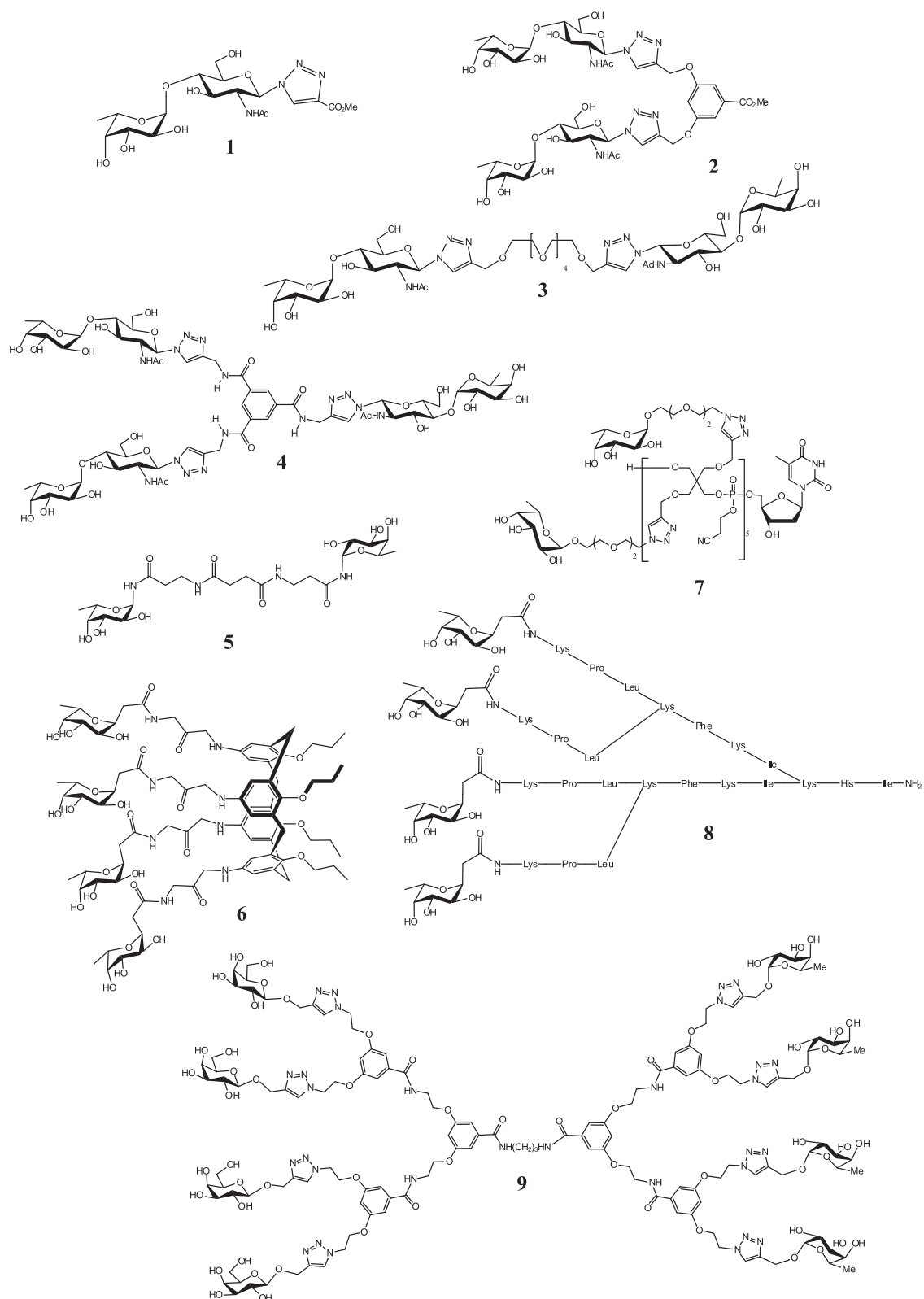


Fig. 5. Selection of glycoconjugates designed and synthesized for their high affinity towards LecB lectin.



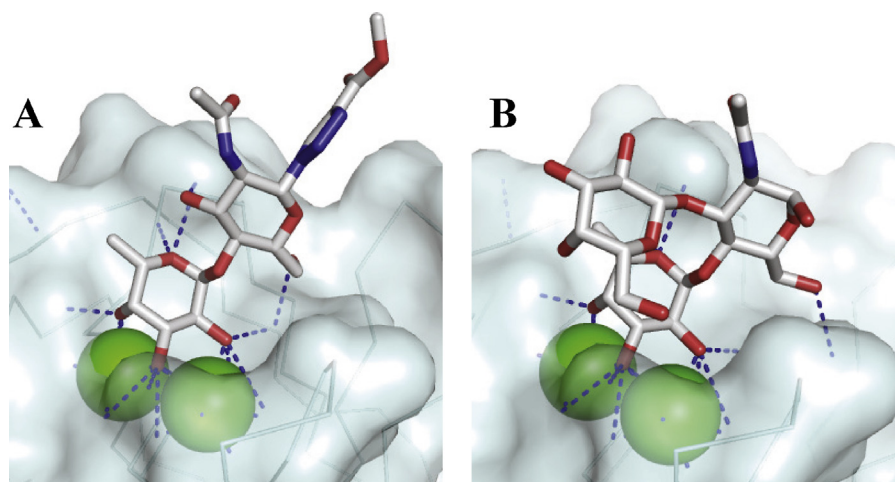


Fig. 6. Comparison of binding sites in the crystal structure of LecB complexed with **(1)** (PDB code 2JDK) [39] and with Lewis a (PDB code 1W8H) [26].

displays reduced cytotoxicity and adhesion on A549 cells compared to parental strain and cause less pronounced lung damages in a murine model of acute pneumonia [36]. In this animal model, co-administration of  $\alpha$ -methyl-fucoside markedly reduced lung injury and mortality. Inhalation of a mixture of fucose and galactose resulted in a decrease of the bacterial load in *P. aeruginosa* pneumonia after treatment of one immune-compromised patient [37], and of a group of cystic fibrosis patients [38].

As a consequence of these promising preliminary data, the search for high affinity ligands for LecB has been initiated by several teams. Since the Lewis a epitope has been identified as the best natural ligand with  $K_d$  of 210 nM [26], a first generation of glycomimetics was designed in order to avoid the laborious synthesis of this trisaccharide [39]. The disaccharide  $\alpha$ Fuc1-4GlcNAc was conserved with addition of several triazole groups on GlcNAc (**1** in Fig. 5) and such compounds were as good ligand as Lewis a (Table 3). Comparison of crystal structures of LecB complexed to **1** and to Lewis a (Fig. 6) demonstrates that the  $\alpha$ Fuc1-4GlcNAc has a slightly different conformation in the two structures. The fucose maintains its coordination with the calcium ions, but the O6 hydroxyl group of the GlcNAc residue establishes different hydrogen bonds. A second generation of compounds containing two such glycomimetics connected through different linkers were then prepared [40]. They were able to bind LecB with the expected stoichiometry and compound **2** and **3** proved to be excellent ligands with  $K_d$  of 170 and 90 nM, respectively. No additional affinity was obtained for the trimeric compound **4**. Attempts to replace the  $\alpha$ Fuc1-4GlcNAc disaccharide by a fucosyl amide also resulted in high affinity ligands for both monovalent and divalent glyco-compounds (**4**) with an additional gain of simple synthesis and chemical and biological stability [41].

In order to reach higher valency, that is often correlated to higher avidity to lectins, other scaffolds have been utilized. A calix[4]arene platform was used to present four  $\alpha$ -L-C-fucosyl units on the same side of the molecule **6** [42].

Glyco-compounds based on propargylated pentaerythrityl phosphodiester oligomers were prepared with four to 10 fucose residues (**7**) [43]. The antenna-like scaffold was preferred by LecB compared to linear or crown-like glyco-clusters synthesized by the same team [44]. Glycopeptide ligands were identified by screening combinatorial libraries of peptide dendrimers [45–47] functionalized with *N*-terminal *C*-fucoside residues at the end of the branches and some tetravalent compounds such as **8** turned out to be potent ligands for LecB [48–50]. Glycodendrimers were also prepared bearing either eight fucose residues, or both galactosides and fucosides (**9**) in order to tackle the two lectins from *P. aeruginosa* [51].

Since the development of therapeutic agents having activity towards bacterial biofilm is of first importance, several of the compounds presented in Fig. 5 have been tested for their ability to inhibit biofilm formation. Compound **6** displayed a dose-dependent biofilm inhibitor effect on *P. aeruginosa* PAO1 strain [42]. Some of the *C*-fucosylpeptide dendrimers such as compound **8** not only

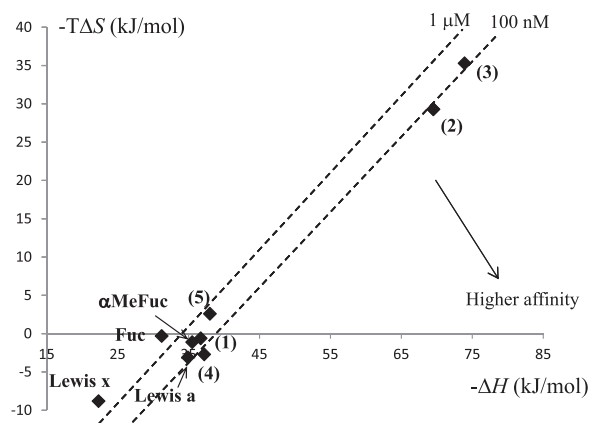


Fig. 7. Enthalpy/entropy plot for the interaction of LecB with natural compounds and with high affinity synthetic glyco-compounds.

**Table 3**

Binding efficiency, competition assays and other activity for a selection of multivalent compounds targeted to LecB.  $K_d$  values are obtained by ITC and  $IC_{50}$  values are obtained by competition against binding to polymeric  $\alpha$ -fucoside (ELLA test).

Compound	$K_d$ ( $\mu$ M)	$IC_{50}$ ( $\mu$ M)	Other activity	Reference
$\alpha$ MeFuc	0.38			[41]
Lewis a	0.21	0.60		[26,39]
<b>1</b>	0.31	0.95		[39]
<b>2</b>	0.17	0.22		[40]
<b>3</b>	0.09	0.35		[40]
<b>4</b>	0.1	0.44		[40]
<b>5</b>	0.62			[41]
<b>6</b>			Inhibition of 80% biofilm formation at 200 $\mu$ M	[42]
<b>7</b>		0.25		[43]
<b>8</b>		0.14	Strong biofilm inhibition	[49]
<b>9</b>			Strong precipitation	[51]

inhibit biofilm formation but are also able to efficiently disperse established biofilms of wild type and hospital strains of *P. aeruginosa* [48,49]. This anti-biofilm effect is of a strong importance for further development of therapeutics to control *P. aeruginosa* infections.

In order to design the next generation of LecB competitors, it is essential to understand the thermodynamic basis of the high affinity obtained with some of the glycomimetics and glycodendrimers described here. The enthalpy-entropy plot displayed in Fig. 7 has been derived from titration microcalorimetry data available for LecB. Some compounds such as bivalent **2** and **3** are clearly highly flexible, resulting in a strong entropy cost in the binding procedure. On the opposite, trivalent compound **4** is sufficiently rigid to avoid the entropy cost but displays a poor enthalpy of binding, indicative of steric limitations. The divalent fucosyl amide compound **5** also displays poor enthalpy of binding, indicating that the NH group at the anomeric position is not as favorable as the anomeric oxygen.

## 7. Perspectives

The design of glycomimetics and glycodendrimers as anti-adhesive compounds for fighting microbial infection is now a very active field of research [52]. Multivalency is an interesting approach for manipulation of protein-carbohydrate interactions and the “glycoside cluster effect” can induce aggregation of several protein receptors, or chelation of several binding sites on the same protein [53–55]. The galactose-specific lectin LecA from *P. aeruginosa* received much attention. The close localization of two of the four sugar binding sites of the tetramer allows the design of glycocompounds with very efficient chelate effect. This results in improvement factor of 800 for tetravalent compounds compared to monovalent ones [56,57]. Clearly, the situation is different in LecB that has a ball shape with fucose-binding sites not oriented in a way that would favor chelating effect. Glycomimetics approach with search for aglycone groups that would enhance the affinity of monovalent fucosylated ligand/or even replacement of fucose by non-carbohydrate group should be envisaged.

On the opposite, the fucose-binding lectins from different *Bukholderia* species present all their carbohydrate-binding sites on the same face of the proteins, i.e.

three sites for BC2L-C-nt and six sites for BamBL as displayed in Fig. 2. These lectins should therefore be good candidate for inhibition by glycoclusters and glycodendrimers.

## Acknowledgments

These studies on bacterial lectins received financial support from CNRS, the Association Vaincre la Mucoviscidose, the GDR Pseudomonas, the COST Actions BM1003 and CM1102 and the Agence Nationale de la Recherche (grants GlycoAsterix 08-PCVI-0028, PA-antiadh 09-JCJC-0047, NeoLect 11-BSV5-002-01 and Labex Arcane ANR-11-LABX-0003).

## References

- [1] I.J. Goldstein, C.E. Hayes, Adv. Carbohydr. Chem. Biochem. 35 (1978) 127.
- [2] H. Lis, N. Sharon, Chem. Rev. 98 (1998) 637.
- [3] H.J. Gabius, S. Andre, H. Kaltner, H.C. Siebert, Biochim. Biophys. Acta 1572 (2002) 165.
- [4] G.R. Vasta, H. Ahmed, E.W. Odom, Curr. Opin. Struct. Biol. 14 (2004) 617.
- [5] M.J. Chrispeels, N.V. Raikhel, Plant Cell 3 (1991) 1.
- [6] H. Kaltner, H.J. Gabius, Adv. Exp. Med. Biol. 491 (2001) 79.
- [7] A. Imberty, A. Varrot, Curr. Opin. Struct. Biol. 18 (2008) 567.
- [8] H.J. Gabius, H.C. Siebert, S. Andre, J. Jimenez-Barbero, H. Rudiger, ChemBiochem 5 (2004) 740.
- [9] N. Sharon, Biochim. Biophys. Acta 28 (2006) 1.
- [10] S.D. Knight, J. Berglund, D. Choudhury, Curr. Opin. Struct. Biol. 4 (2000) 653.
- [11] E.A. Merritt, W.G.J. Hol, Curr. Opin. Struct. Biol. 5 (1995) 165.
- [12] A. Imberty, Bacterial lectins and adhesins: structures, ligands and functions, in : O. Renaudet, N. Spinelli (Eds.), E-book: Synthesis and Biological Applications of Glycoconjugates, Bentham Science Publishers Ltd, Bussum, The Netherlands, 2011, p. 3.
- [13] J.R. Bishop, P. Gagneux, Glycobiology 17 (2007) 23R.
- [14] C. Perrier, N. Sprenger, B. Corthesy, J. Biol. Chem. 281 (2006) 14280.
- [15] W.M. Watkins, W.T. Morgan, Nature 169 (1952) 825.
- [16] S.A. Berger, N.A. Young, S.C. Edberg, Eur. J. Clin. Microbiol. Infect. Dis. 8 (1989) 681.
- [17] A.E. Mourant, A.C. Kopec, Blood groups and diseases, Oxford University Press, Oxford, 1978.
- [18] S. Holmner, A. Mackenzie, U. Krengel, FEBS Lett. 584 (2010) 2548.
- [19] P.K. Mandal, T.R. Branson, E.D. Hayes, J.F. Ross, J.A. Gavin, A.H. Daranas, W.B. Turnbull, Angew. Chem. Int. Ed. Engl. 51 (2012) 5143.
- [20] I. Vigan-Womas, M. Guillotte, A. Juillerat, A. Hessel, B. Raynal, P. England, J.H. Cohen, O. Bertrand, T. Peyrard, G.A. Bentley, A. Lewit-Bentley, O. Mercereau-Puijalon, PLoS Pathog. 8 (2012) e1002781.
- [21] L. Lindesmith, C. Moe, S. Marionneau, N. Ruvoen, X. Jiang, L. Lindblad, P. Stewart, J. LePendou, R. Baric, Nat. Med. 9 (2003) 548.
- [22] E. Mitchell, C. Houles, D. Sudakevitz, M. Wimmerova, C. Gautier, S. Pérez, A.M. Wu, N. Gilboa-Garber, A. Imberty, Nat. Struct. Biol. 9 (2002) 918.



- [23] E.P. Mitchell, C. Sabin, L. Šnajdrová, M. Pokorná, S. Perret, C. Gautier, C. Hofr, N. Gilboa-Garber, J. Koča, M. Wimmerová, A. Imberty, *Proteins Struct. Funct. Bioinfo.* 58 (2005) 735.
- [24] M. Pokorná, G. Cioci, S. Perret, E. Rebuffet, N. Kostlánová, J. Adam, N. Gilboa-Garber, E.P. Mitchell, A. Imberty, M. Wimmerová, *Biochemistry (Mosc)* 45 (2006) 7501.
- [25] A. Audfray, J. Claudinon, S. Abounit, N. Ruvoën-Clouet, G. Larson, D.F. Smith, M. Wimmerová, J. Le Pendu, W. Römer, A. Varrot, A. Imberty, *J. Biol. Chem.* 287 (2012) 4335.
- [26] S. Perret, C. Sabin, C. Dumon, M. Pokorná, C. Gautier, O. Galanina, S. Ilia, N. Bovin, M. Nicaise, M. Desmadril, N. Gilboa-Garber, M. Wimmerová, E.P. Mitchell, A. Imberty, *Biochem. J.* 389 (2005) 325.
- [27] O. Šulák, G. Cioci, M. Delia, M. Lahmann, A. Varrot, A. Imberty, M. Wimmerová, *Structure* 18 (2010) 59.
- [28] O. Šulák, G. Cioci, E. Lameignère, V. Balloy, A. Round, I. Gutsche, L. Malinová, M. Chignard, P. Kosma, F. Aubert, C.L. Marolda, M.A. Valvano, M. Wimmerová, A. Imberty, *PLoS Pathog.* 7 (2011) e1002238.
- [29] C. Sabin, E.P. Mitchell, M. Pokorná, C. Gautier, J.P. Utille, M. Wimmerová, A. Imberty, *FEBS Lett.* 580 (2006) 982.
- [30] T.K. Dam, C.F. Brewer, *Chem. Rev.* 102 (2002) 387.
- [31] R. Loris, D. Tielker, K.E. Jaeger, L. Wyns, *J. Mol. Biol.* 331 (2003) 861.
- [32] A. Imberty, E.P. Mitchell, M. Wimmerová, *Curr. Opin. Struct. Biol.* 15 (2005) 525.
- [33] D. Tielker, S. Hacker, R. Loris, M. Strathmann, J. Wingender, S. Wilhelm, F. Rosenau, K.E. Jaeger, *Microbiology* 151 (2005) 1313.
- [34] R. Marchetti, L. Malinová, E. Lameignère, L. Adamová, C. de Castro, G. Cioci, C. Stanetty, P. Kosma, A. Molinaro, M. Wimmerová, A. Imberty, A. Silipo, *Glycobiology* 22 (2012) 1387.
- [35] H. Gustke, R. Kleene, G. Loers, N. Nehmann, M. Jaehne, K.M. Bartels, K.E. Jaeger, M. Schachner, U. Schumacher, *Eur. J. Clin. Microbiol. Infect. Dis.* 31 (2012) 207.
- [36] C. Chemani, A. Imberty, S. de Bentzman, P. Pierre, M. Wimmerová, B.P. Guery, K. Faure, *Infect. Immun.* 77 (2009) 2065.
- [37] P. von Bismarck, R. Schneppenheim, U. Schumacher, *Klin. Padiatr.* 213 (2001) 285.
- [38] H.P. Hauber, M. Schulz, A. Pforte, D. Mack, P. Zabel, U. Schumacher, *Int. J. Med. Sci.* 5 (2008) 371.
- [39] K. Marotte, C. Sabin, C. Preville, M. Moume-Pymbock, M. Wimmerová, E.P. Mitchell, A. Imberty, R. Roy, *ChemMedChem* 2 (2007) 1328.
- [40] K. Marotte, C. Preville, C. Sabin, M. Moumé-Pymbock, A. Imberty, R. Roy, *Org. Biomol. Chem.* 5 (2007) 2953.
- [41] M. Andreini, M. Anderluh, A. Audfray, A. Bernardi, A. Imberty, *Carbohydr. Res.* 345 (2010) 1400.
- [42] G.M.L. Consoli, G. Granata, V. Cafiso, S. Stefani, C. Geraci, *Tetrahedron Lett.* 52 (2011) 5831.
- [43] F. Morvan, A. Meyer, A. Jochum, C. Sabin, Y. Chevolut, A. Imberty, J.P. Praly, J.J. Vasseur, E. Souteyrand, S. Vidal, *Bioconjug. Chem.* 18 (2007) 1637.
- [44] B. Gerland, A. Goudot, G. Pourceau, A. Meyer, V. Dugas, S. Cecioni, S. Vidal, E. Souteyrand, J.J. Vasseur, Y. Chevolut, F. Morvan, *Bioconjug. Chem.* 23 (2012) 1534.
- [45] A. Clouet, T. Darbre, J.L. Reymond, *Angew. Chem., Int. Ed. Engl.* 43 (2004) 4612.
- [46] N. Maillard, A. Clouet, T. Darbre, J.L. Reymond, *Nat. Protoc.* 4 (2009) 132.
- [47] J.L. Reymond, T. Darbre, *Org. Biomol. Chem.* 10 (2012) 1483.
- [48] E.M. Johansson, S.A. Cruz, E. Kolomiets, L. Buts, R.U. Kadam, M. Cacciarini, K.M. Bartels, S.P. Diggle, M. Camara, P. Williams, R. Loris, C. Nativi, F. Rosenau, K.E. Jaeger, T. Darbre, J.L. Reymond, *Chem. Biol.* 15 (2008) 1249.
- [49] E.M.V. Johansson, R.U. Kadam, G. Rispoli, S.A. Cruz, K.M. Bartels, S.P. Diggle, M. Camara, P. Williams, K.E. Jaeger, T. Darbre, J.L. Reymond, *MedChemComm* 2 (2011) 418.
- [50] E. Kolomiets, M.A. Swiderska, R.U. Kadam, E.M. Johansson, K.E. Jaeger, T. Darbre, J.L. Reymond, *ChemMedChem* 4 (2009) 562.
- [51] I. Deguise, D. Lagnoux, R. Roy, *New J. Chem.* 31 (2007) 1321.
- [52] A. Imberty, Y.M. Chabre, R. Roy, *Chem. Eur. J.* 14 (2008) 7490.
- [53] Y.M. Chabre, R. Roy, *Adv. Carbohydr. Chem. Biochem.* 63 (2010) 165.
- [54] R.J. Pieters, *Org. Biomol. Chem.* 7 (2009) 2013.
- [55] M. Reynolds, S. Perez, *C. R. Chimie* 14 (2011) 74.
- [56] S. Cecioni, R. Lalor, B. Blanchard, J.P. Praly, A. Imberty, S.E. Matthews, S. Vidal, *Chem. Eur. J.* 15 (2009) 13232.
- [57] S. Cecioni, J.P. Praly, S.E. Matthews, M. Wimmerová, A. Imberty, S. Vidal, *Chem. Eur. J.* 18 (2012) 6250.
- [58] B. Blanchard, A. Nurisso, E. Hollville, C. Tétaud, J. Wiels, M. Pokorná, M. Wimmerová, A. Varrot, A. Imberty, *J. Mol. Biol.* 383 (2008) 837.
- [59] E. Lameignère, T.C. Shiao, R. Roy, M. Wimmerová, F. Dubreuil, A. Varrot, A. Imberty, *Glycobiology* 20 (2010) 87.